

**NONCATALEPTIC NEUROLEPTIC AGENTS:
SYNTHESIS OF SOME ESTERS OF 2-CHLORO-10-
-(4-(2-HYDROXYETHYL)PIPERAZINO)-10,11-
-DIHYDRODIBENZO[*b,f*]THIEPIN**

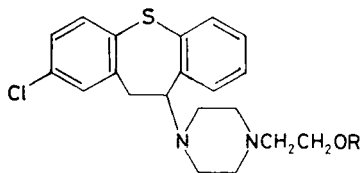
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Reactions of 2-chloro-10-(4-(2-hydroxyethyl)piperazino)-10,11-dihydrodibenzo[*b,f*]thiepin (*I*) with phenylacetic, methoxyacetic, methylthioacetic, phenoxyacetic and morpholinoacetic acid in dichloromethane and in the presence of *N,N'*-carbonyldiimidazole gave the title esters *II–VI*. Reaction of *I* with succinic anhydride afforded the hemisuccinate *VII*. The esters prepared elicited ataxia in low doses, were low-cataleptic, but only *II, IV, and VII* proved some antidopaminergic activity in the test using the affecting dopamine metabolism in rat brain striatum.

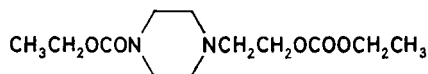
2-Chloro-10-(4-(2-hydroxyethyl)piperazino)-10,11-dihydrodibenzo[*b,f*]thiepin ("docloxythiepin", *I*) (refs^{1–3}) showed in animal and biochemical tests^{5,6} properties of a potentially useful noncataleptic neuroleptic agent. A pharmacokinetic study⁷ of the radioactively labelled compound ($10\text{-}^{14}\text{C-I}$) in rats, administered orally in the form of the well water-soluble succinate, proved a high-rate elimination of radioactivity in faeces (80% in four days) and indicated thus a low absorption. This indication was partly contradicted by finding high docloxythiepin (*I*) levels in bile and an evidently intensive enterohepatic cycle⁸. In the meantime, however, there was attempted to modulate the bioavailability of docloxythiepin (*I*) by its substitution with some esters as potential pro-drugs. The choice of these esters was empirical



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|-----------------------------------------------------------------|-----------------------------------------------------------------------------------------|
| <i>I</i> , R = H | <i>V</i> , R = COCH ₂ OC ₆ H ₅ |
| <i>II</i> , R = COCH ₂ C ₆ H ₅ | <i>VI</i> , R = COCH ₂ N $\begin{array}{c} \diagup \\ \diagdown \end{array}$ |
| <i>III</i> , R = COCH ₂ OCH ₃ | <i>VII</i> , R = CO(CH ₂) ₂ COOH |
| <i>IV</i> , R = COCH ₂ SCH ₃ | |

but at least partly it was tried to maintain the hydrophobic-hydrophilic balance similar like in the docloxythepin (*I*) molecule; some rather hydrophilic acids were therefore used for esterification. The synthesis and pharmacology of esters *II–VII* is being described in this communication.

Out of the docloxythepin esters only acetate² and decanoate¹ are known. Reactions of phenylacetic, methoxyacetic⁹, methylthioacetic¹⁰, phenoxyacetic and morpholinoacetic acid¹¹ with *N,N'*-carbonyldiimidazole in dichloromethane and the following treatment with the base *I* at room temperature afforded the esters *II–VII* (for the method, *cf.*¹²). With the exception of the ester *III*, the bases were oily. They were transformed to crystalline maleates and characterized by the mass spectra (the base *III* by IR and ¹H NMR spectra). Reaction of the base *I* with succinic anhydride in boiling benzene gave the monobasic hemisuccinate *VII*, identified by spectra and transformed to the hydrogen maleate. An attempt to prepare docloxythepin monoethyl carbonate by treatment of the base *I* with an equivalent amount of ethyl chloroformate in benzene at 35°C led to a cleavage of the molecule of *I*. The product, which crystallized from the reaction mixture, was identified as the monohydrochloride of *I*. Processing of the mother liquor with maleic acid gave a crystalline maleate which was identified as 1-(ethoxycarbonyl)-4-(2-ethoxycarbonyloxyethyl)-piperazine (*VIII*) hydrogen maleate (analysis and mass spectrum). Esterification of compound *I* with ethyl chloroformate proceeds under simultaneous cleavage of the C₍₁₀₎—N₍₁₎ bond and N₍₁₎—acylation. A part of the starting *I* (50% in fact) remains unchanged and binds a part of hydrogen chloride formed to give the monohydrochloride. The neutral product of the cleavage reaction was evidently 2,10-dichloro-10, 11-dihydrodibenzo[*b,f*]thiepin¹ (*cf.*¹³) which was not isolated.



VIII

Results of preliminary pharmacological testing of the compounds prepared (compounds *II–VII* were tested in the form of maleates, administered orally; the doses given were calculated per bases): Acute toxicity in mice, LD₅₀ in mg/kg: *II*, 240; *III*, 220; *V*, 326; *VI*, 170; *VII*, 150. Discoordinating activity in the rotarod test in mice, ED₅₀ in mg/kg: *II*, 1.0; *III*, 3.1; *IV*, 3.0; *V*, 1.9; *VI*, 2.0; *VII*, 1.1. Cataleptic effect in rats, percentage of cataleptic animals after the dose of 100 mg/kg: *II*, 10; *III*, 30; *IV*, 30; *V*, 50 (= ED₅₀); *VI*, 30; *VII*, 30 (the cataleptogenic effects of the whole series are negligible). Antiapomorphine effect in rats (antagonization of the stereotypies): compounds *II–VII* inactive at the dose of 50 mg/kg. Antidopaminergic effects evaluated by determination of levels of homovanillic (HVA) and 3,4-dihydroxyphenyl acetic acid (DOPAC) (main dopamine metabolites) in rat brain striatum

in comparison with the control: compounds *II* and *VII* in doses of 80 mg/kg brought about rises of HVA levels by 150–200%. The same doses of compounds *III*, *V*, and *VI* were practically without effect. Compound *IV* increased significantly the HVA level at the dose of 40 mg/kg and the DOPAC level at the dose of 20 mg/kg (active in similar doses like docloxythepin⁶). The results were not considered warranting a pharmacokinetic study of any of the esters.

Compounds *II–VII* were also tested for antimicrobial activity *in vitro* (microorganisms and the minimum inhibitory concentrations in µg/ml given unless they exceed 100 µg/ml): *Streptococcus β-haemolyticus*, *III* 50, *V* 50, *VI* 25; *Streptococcus faecalis*, *III* 50, *V* 50, *VI* 100; *Staphylococcus pyogenes aureus*, *III* 25, *V* 25, *VII* 100; *Escherichia coli*, *III* 50, *V* 12.5, *VI* 25; *VII* 100; *Proteus vulgaris*, *V* 100, *VI* 100; *Mycobacterium tuberculosis* H37Rv, *III* 12.5, *VI* 12.5; *Trichophyton mentagrophytes*, *II* 50, *III* 50, *IV* 50, *V* 50.

EXPERIMENTAL

The melting points of analytical samples were determined in a Mettler FP-5 melting point recorder; the samples were dried *in vacuo* of about 60 Pa over P₂O₅ at room temperature or at 77°C. IR spectra (in Nujol) were recorded with a Unicam SP 200G spectrophotometer, ¹H NMR spectra (in C²HCl₃) with a Tesla BS 487 C (80 MHz) spectrometer, and mass spectra with the MCH-1320 spectrometer. The homogeneity of the products and composition of the mixtures were checked by thin-layer chromatography on silica gel (Silufol).

2-Chloro-10-(4-(2-phenylacetoxyethyl)piperazino)-10,11-dihydrodibenzo[*b,f*]thiepin (*II*)

A solution of 3.20 g 90% N,N'-carbonyldiimidazole in 40 ml dichloromethane was stirred under nitrogen and treated with 2.45 g phenylacetic acid. After 10 min stirring 6.0 g *I* (ref.¹) were added, the mixture was stirred for 2 h at room temperature, allowed to stand for 48 h, and evaporated *in vacuo*. The residue was dissolved in 50 ml benzene, precipitated imidazole was filtered off after 2 h and washed with benzene. The filtrate was evaporated *in vacuo*, the glassy residue was dissolved in 20 ml acetone and the solution was treated with a solution of 1.80 g maleic acid in 10 ml acetone. It was followed by crystallization of 0.80 g substance, m.p. 136–137°C with decomposition (needles from ethanol), which was identified as imidazole hydrogen maleate. For C₇H₈N₂O₄ (184.2) calculated: 45.65% C, 4.38% H, 15.22% N; found: 46.07% C, 4.45% H, 15.43% N.

The mother liquor was partly evaporated *in vacuo* and by standing for 3 h the maleate of *II* crystallized; 5.20 g (54%), m.p. 154–155°C (acetone). Mass spectrum (*m/z*): 492 (M⁺ corresponding to C₂₈H₂₉ClN₂O₂S), 356, 245, 163, 91. For C₃₂H₃₃ClN₂O₆S (609.2) calculated: 63.10% C, 5.46% H, 5.82% Cl, 4.60% N, 5.26% S; found: 62.60% C, 5.65% H, 6.03% Cl, 4.70% N, 5.40% S.

2-Chloro-10-(4-(2-methoxyacetoxyethyl)piperazino)-10,11-dihydrodibenzo[*b,f*]thiepin (*III*)

A stirred solution of 3.25 g 90% N,N'-carbonyldiimidazole in 30 ml dichloromethane was treated under nitrogen with a solution of 1.80 g methoxyacetic acid (b.p. 97–99°C/2 kPa) (ref.⁹) in 4 ml

dichloromethane. It was stirred for 10 min, treated with a solution of 6.0 g *I* (ref.¹) in 10 ml dichloromethane, stirred for 2 h at room temperature, allowed to stand for 48 h, and evaporated *in vacuo*. The residue was dissolved in 50 ml benzene, the solution deposited crystalline imidazole which was filtered off, the filtrate was evaporated and the residue crystallized by standing; 5.78 g (81%), m.p. 103–104°C (acetone). IR spectrum: 760, 819, 875 (4 and 2 adjacent and solitary Ar—H), 1 137, 1 200 (C—O of ether and ester), 1 560, 1 583, 3 038, 3 060 (Ar), 1 750 cm⁻¹ (RCOOR'). ¹H NMR spectrum: δ 6.90–7.70 (m, 7 H, ArH), 4.25 (t, *J* = 7.0 Hz, 2 H, CH₂OCO), 4.00 (s, 2 H, COCH₂O), 3.00–4.00 (m, 3 H, ArCH₂CHAr), 3.40 (s, 3 H, CH₃O), 2.40–2.80 (m, 10 H, 5 CH₂N). For C₂₃H₂₇ClN₂O₃S (447.0) calculated: 61.80% C, 6.09% H, 7.93% Cl, 6.27% N, 7.17% S; found: 62.27% C, 6.01% H, 7.92% Cl, 6.19% N, 7.48% S.

Maleate, m.p. 153–154°C (acetone). For C₂₇H₃₁ClN₂O₇S (563.1) calculated: 57.59% C, 5.55% H, 6.30% Cl, 4.98% N, 5.69% S; found: 58.00% C, 5.48% H, 6.31% Cl, 5.24% N, 5.90% S.

2-Chloro-10-(4-(2-methylthioacetoxyethyl)piperazino)-
-10,11-dihydrodibenzo[*b,f*]thiepin (*IV*)

A similar reaction of 3.25 g 93% N,N'-carbonyldiimidazole with 1.7 g methylthioacetic acid (b.p. 108°C/1.3 kPa) (ref.¹⁰) in 70 ml dichloromethane, treatment with 6.0 g *I* (ref.¹) in 15 ml dichloromethane, and similar processing gave 6.1 g oily product containing some imidazole. It was chromatographed on 90 g neutral Al₂O₃ (activity II). Elution with benzene afforded 5.5 g (74%) homogeneous oily *IV*. Neutralization with maleic acid in acetone and addition of ether gave the maleate, m.p. 137–137.5°C (acetone-ether). Mass spectrum, *m/z*: 462 and 464 (M⁺, corresponding to C₂₃H₂₇ClN₂O₂S₂), 356, 357, 245 (base peak). For C₂₇H₃₁ClN₂O₆S₂ (579.1) calculated: 56.00% C, 5.39% H, 6.12% Cl, 4.84% N, 11.07% S; found: 56.00% C, 5.40% H, 6.22% Cl, 4.74% N, 10.88% S.

2-Chloro-10-(4-(2-phenoxyacetoxyethyl)piperazino)-
-10,11-dihydrodibenzo[*b,f*]thiepin (*V*)

Similarly like in the case of *II*, 3.2 g N,N'-carbonyldiimidazole were treated first with 2.75 g phenoxyacetic acid and then with 6.0 g *I* (ref.¹) in 40 ml dichloromethane. Similar processing including removal of imidazole gave 8.0 g crude oily product which was neutralized with 1.90 g maleic acid in 40 ml acetone. A small quantity of imidazole hydrogen maleate was filtered off and the filtrate gave by crystallization 7.30 g (72%) *V* maleate hemihydrate, m.p. 150–152°C (acetone). For C₃₂H₃₃ClN₂O₇S + 0.5 H₂O (634.1) calculated: 60.60% C, 5.40% H, 5.59% Cl, 4.42% N, 5.06% S; found: 60.91% C, 5.36% H, 6.00% Cl, 4.52% N, 5.16% S.

2-Chloro-10-(4-(2-morpholinoacetoxyethyl)piperazino)-
-10,11-dihydrodibenzo[*b,f*]thiepin (*VI*)

N,N'-Carbonyldiimidazole (3.25 g, 93%) was similarly treated with 2.32 g morpholinoacetic acid (m.p. 162–163°C) (ref.¹¹) in 70 ml dichloromethane and after 15 min of stirring with a solution of 6.0 g *I* (ref.¹) in 15 ml dichloromethane. The mixture was stirred for 2 h at room temperature, allowed to stand for 20 h and evaporated. The residue was dissolved in 50 ml benzene and the solution allowed to stand overnight. The crystallized imidazole was filtered off and the filtrate was evaporated *in vacuo*. The residue was dissolved in 70 ml acetone and the solution was neutralized with 3.8 g maleic acid in 12 ml acetone. The resulting solution was partly evaporated and diluted with ether; 8.20 g (69%) *VI* dimaleate hemihydrate, m.p. 132.5–133.5°C (acetone-ether). Mass spectrum, *m/z*: 501 (M⁺ corresponding to C₂₆H₃₂ClN₃O₃S), 245 (100%), 210, 144, 113

100 (100%), 84, 70, 55. For $C_{34}H_{40}ClN_3O_{11}S + 0.5 H_2O$ (743.2) calculated: 54.94% C, 5.56% H, 4.77% Cl, 5.65% N, 4.32% S; found: 54.92% C, 5.50% H, 5.13% Cl, 5.56% N, 4.64% S.

2-Chloro-10-(4-(2-(3-carboxypropionyloxy)ethyl)piperazino)-10,11-dihydrodibenzo[*b,f*]thiepin (*VII*)

A solution of 6.0 g *I* (ref.¹) in 75 ml benzene was treated with 1.6 g succinic anhydride and the mixture was stirred and refluxed for 5 h. After evaporation of benzene *in vacuo*, the residue was triturated with a mixture of 20 ml light petroleum and 20 ml acetone and allowed to stand overnight; 6.6 g (87%) *VII*, m.p. 149°C (acetone). IR spectrum: 749, 821, 890 (4 and 2 adjacent and solitary Ar—H), 1 175, 1 360, 1 380 (C—O), 1 563, 1 580, 3 030, 3 050 (Ar), 980, 1 700 (COOH), 1 749 cm^{-1} (RCOOR'). ¹H NMR spectrum: δ 11.68 (bs, 1 H, COOH), 6.90—7.70 (m, 7 H, ArH), 4.25 (bt, 2 H, CH₂OCO), 3.00—4.00 (m, 3 H, ArCH₂CHAR), 2.70 (bs, 10 H, 5 CH₂N), 2.54 (s, 4 H, COCH₂CH₂CO). For $C_{24}H_{27}ClN_2O_4S$ (475.0) calculated: 60.69% C, 5.73% H, 7.46% Cl, 5.90% N, 6.75% S; found: 61.14% C, 6.07% H, 7.27% Cl, 5.92% N, 6.78% S.

Maleate, m.p. 141.5—142°C (acetone-ether). For $C_{28}H_{31}ClN_2O_8S$ (591.1) calculated: 56.90% C, 5.29% H, 6.00% Cl, 4.74% N, 5.42% S; found: 57.20% C, 4.75% H, 6.11% Cl, 4.70% N, 5.61% S.

1-(Ethoxycarbonyl)-4-(2-ethoxycarbonyloxyethyl)piperazine (*VIII*)

A stirred solution of 6.0 g *I* (ref.¹) in 30 ml benzene was treated dropwise with a solution of 1.80 g ethyl chloroformate in 10 ml benzene at 35°C. The mixture was allowed to stand for 72 h at room temperature and the precipitate was filtered, washed with benzene and crystallized from a mixture of aqueous ethanol and acetone; 3.3 g (50%) *I* monohydrochloride, m.p. 220—223°C. Mass spectrum, *m/z*: 374 (M^+ corresponding to $C_{20}H_{23}ClN_2OS$), 343 ($M-CH_2OH$), 273, 272, 245, 129, 100 (base peak). For $C_{20}H_{24}Cl_2N_2OS$ (411.4) calculated: 58.39% C, 5.88% H, 17.24% Cl, 6.81% N, 7.79% S; found: 59.02% C, 5.86% H, 16.87% Cl, 6.63% N, 7.70% S.

The mother liquor was evaporated *in vacuo*, the residue was dissolved in 5 ml acetone and the solution was treated with 1.0 g maleic acid in 2 ml acetone. Crystallization by standing gave 3.85 g mixture melting at 122—123°C. Repeated crystallization from acetone gave finally 1.08 g (17%) *VIII* hydrogen maleate, m.p. 140—141°C. Mass spectrum (*m/z*): 274 (M^+ corresponding to $C_{12}H_{22}N_2O_5$), 229, 185, 184, 171 (base peak). For $C_{16}H_{29}N_2O_9$ (390.4) calculated: 49.22% C, 6.71% H, 7.17% N, found: 49.19% C, 6.78% H, 7.25% N.

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